

# Chapter 22

## Metal Oxide Based Biosensors for the Detection of Dangerous Biological Compounds

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**Abstract** In this report an application of some metal oxide nanostructures as a biosensor platform for the detection of dangerous biological compounds (Bovine leucosis, Salmonella) have been discussed. The attention is paid to the TiO<sub>2</sub> nanoparticles and ZnO nanorods deposited on the flat surface. The changes in photoluminescence signal from nanostructured surface were applied as biosensor response to detect the analytes. The detection range of TiO<sub>2</sub> based biosensor for Bovine leucosis antibodies was in the range of 2–10 μg/ml. The detection range of ZnO based biosensor for Salmonella antigens was 10<sup>2</sup>–10<sup>6</sup> cells/ml. The obtained results provide a good basis for the use of optical properties of metal oxide based semiconductor nanostructures in biosensor technology.

### 22.1 Introduction

During last decades metal oxide nanostructures based on TiO<sub>2</sub> and ZnO are the materials that attract a lot of attention due to their optical, catalytic and sensing applications. Physico-chemical properties of these nanomaterials that can be controlled and changed by growth methods or by modification of nanostructures can play crucial role in sensing application. For these metal oxides, quantum confinement effect caused by nanoscale size, have resulted not only in band gap increase and improved photocatalytic activity but also in photoluminescence peaks appearance at room temperature. Being wide band gap semiconductors that have a good affinity to biological compounds, TiO<sub>2</sub> and ZnO nanostructures are promising materials to be used as optical biosensor transducers [1–5].

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Among various types of biosensors, an immune biosensor is a type, based on specific interaction between antibody (Ab) – antigen (Ag) couple [6]. The reaction between this couples has high specificity and sensitivity to detecting analytes what makes immune biosensors suitable for accurate and precise tests with electrochemical, optical, magnetic and piezoelectric transducers [6–8]. As it is known from the range of works, optical methods of detection based on absorbance, reflectance and photoluminescence demonstrate simple, fast and accurate detection of target analyte [9–14]. In particular, photoluminescence from nanostructured metal oxides is a promising property that can be used for the detection of chemical and biologic compounds [4, 9, 10, 13, 14]. Therefore, the photoluminescence from TiO<sub>2</sub> nanoparticles was used for the detection of Bovine leucosis antibodies.

### ***22.1.1 Application of Photoluminescence TiO<sub>2</sub> Nanoparticles for the Detection of Bovine Leucosis***

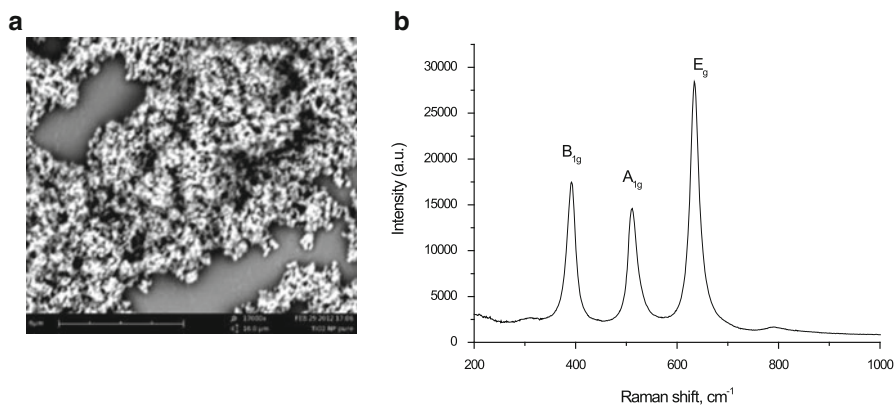
Bovine leucosis virus (BLV) – is the highly foetal neoplasia of the cattle characterized by the abnormality maturation process of the blood cells [15]. Diagnosis of the BLV infection based on the clinical signs alone is difficult because of the wide range of symptoms. The traditional immune methods have high specificity and sensitivity, but they take a lot of time and require additional parameters such as labelled molecules [16]. To overcome such drawbacks we need to use the modern instrumental analytical devices based on the biosensor technology.

TiO<sub>2</sub> is a material, which is widely applied for different applications [17–22] including sensors and biosensors [20, 23–25]. TiO<sub>2</sub> shows good stability in aggressive environment what makes it attractive for chemical sensors applications [18, 20]. TiO<sub>2</sub> is a wide band gap semiconductor with indirect optical transitions [19]. TiO<sub>2</sub> has low isoelectric point pH = 5.5 what makes some advantages to protein immobilization on its surface [25].

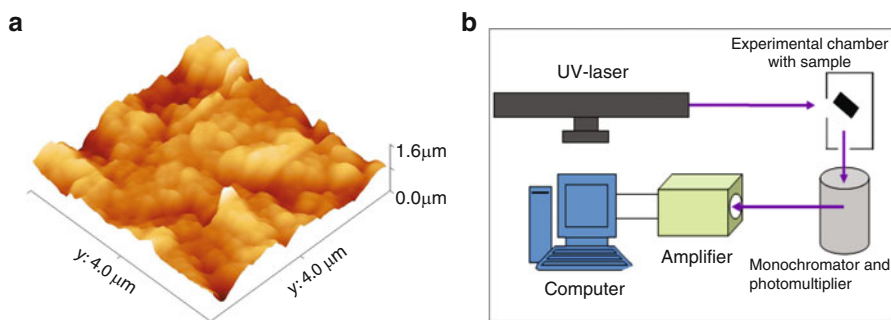
## **22.2 Experimental**

Anatase nanoparticles with mean size 32 nm were used as biosensor template. TiO<sub>2</sub> nanoparticles were solved in water to prepare sols (with concentration 0.05 mg/ml). TiO<sub>2</sub> layers were formed on glass substrates by dropping TiO<sub>2</sub> sols and drying at room temperature with post annealing treatment at 300 C for 1 h was provided to remove water from the samples. SEM measurements showed that nanoparticles formed high surface area porous structure (Fig. 22.1a) that is suitable platform for immobilization of biological species.

Raman spectrometer with Ar/Kr laser (Jobin Yvon-Labram 1B,  $\lambda = 647.1$  nm) and spectral resolution 1 cm<sup>-1</sup> were used for the study of Raman spectra. Raman spectrum of TiO<sub>2</sub> nanostructures, deposited on glass substrates is shown in Fig. 22.1b. The peaks were found at 392, 512 and 634 cm<sup>-1</sup>, which correspond to B<sub>1g</sub>, A<sub>1g</sub> and E<sub>g</sub> modes of anatase phase of TiO<sub>2</sub>. Surface morphology of



**Fig. 22.1** (a) – SEM image of TiO<sub>2</sub> nanostructures deposited on glass; (b) – Raman spectrum of TiO<sub>2</sub> nanostructures

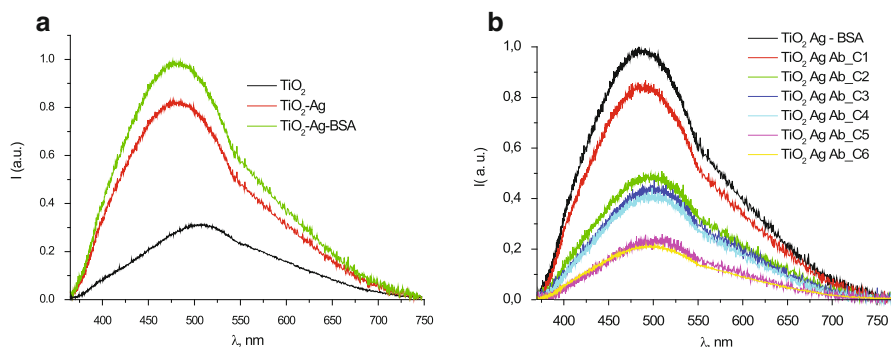


**Fig. 22.2** (a) – AFM image of surface of TiO<sub>2</sub> nanostructures (Asylum Research MFP-3D); (b) – photoluminescence setup

deposited samples has been investigated with AFM (Fig. 22.2a). The obtained TiO<sub>2</sub> nanostructures had high active surface area. Mean square surface roughness (Rsq), measured with free software Gwiddion, was 140 nm for prepared TiO<sub>2</sub> nanostructures. Photoluminescence spectra were measured by setup shown on Fig. 22.2b. The photoluminescence was stimulated by UV laser LCS-DTL-374QT with excitation wavelength  $\lambda = 355$  nm. The emission spectra were amplified and recorded in the range of 370–800 nm.

## 22.3 Results and Discussion

To fabricate biosensitive layer, the antigens of BLV were immobilized on TiO<sub>2</sub> surface. TiO<sub>2</sub> nanostructures were exposed to water solution of BLV antigens (Ag) for 10 min and then were washed two times in distilled water and dried in air at room temperature. The backside of TiO<sub>2</sub> sample was sealed to prevent immobilization of Ag on it.



**Fig. 22.3** (a) – PL spectra of  $\text{TiO}_2$  nanoparticles before and after immobilization of BLV antigens, and after BSA adsorption; (b) – PL spectra of  $\text{TiO}_2$ -Ag-BSA layer under different concentrations of BLV Ab

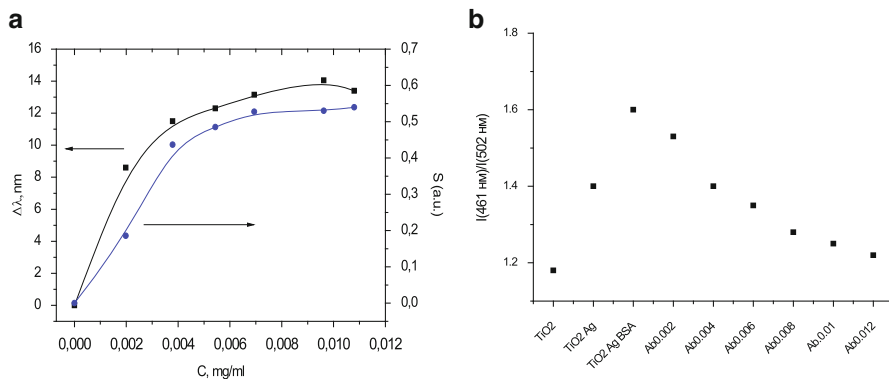
The photoluminescence spectrum of pure  $\text{TiO}_2$  samples is characterized by broad peak centered at 510 nm. A number of papers reported on room temperature PL in  $\text{TiO}_2$  nanostructures [22, 26–28]. Usually  $\text{TiO}_2$  nanostructures demonstrate emission in the range 430–560 nm. Two mechanisms of luminescence are proposed: self-trapped excitons (STE) (430–510 nm) and oxygen vacancies (530–560 nm).

Photoluminescence (PL) spectra of  $\text{TiO}_2$  nanoparticles before and after immobilization of antigens are shown on Fig. 22.3a. As one can see, the immobilization of BLV antigens lead to the significant changes of PL spectra in intensity and peak position. It was found that after Ag immobilization PL spectra was shifted to shorter wavelengths, what can be a proof of formation links between  $\text{TiO}_2$  and Ag. Increase of PL intensity could result from charge transfer between Ag molecules and conductance band of  $\text{TiO}_2$ . And the UV shift of PL maximum can be explained by additional dipole-dipole interaction, what can change energetic position of recombination centers into  $\text{TiO}_2$ . After immobilization of antigens, the bovine serum albumin (BSA) was deposited on the biosensitive layer as a blocking agent to prevent nonspecific protein adsorption. The intensity of PL after BSA adsorption has been increased.

After forming of biosensitive layer by immobilization BLV antigens, the BLV antibodies (which play role of analyte) were deposited on the functionalized surface from water solutions with different concentrations. PL spectra of  $\text{TiO}_2$ -Ag-BSA biosensor, measured under different Ab concentrations are shown in Fig. 22.3b. It was found that PL intensity decreased with the increase of analyte concentration. At the same time, peak position moved to higher wavelengths.

Thus, the biosensor response to leucosis Ab can be a function of two parameters: PL intensity and position of PL peak. To analyze the sensor response we calculated the changes biosensor signal  $S$  according the following equation

$$S = \frac{S_{\text{Ag-BSA}} - S_{\text{Ab}}}{S_{\text{Ag-BSA}}}, \quad (22.1)$$



**Fig. 22.4** (a) – Response of the biosensor signal  $S$  and peak shift to different concentrations of BLV antibodies; (b) Ratio of  $I_{\text{STE}}/I_{V[\text{O}]}$  for  $\text{TiO}_2$  NP before and after interaction with biomolecules

where  $S_{\text{Ag-BSA}}$  and  $S_{\text{Ab}}$  are PL peak's intensities of  $\text{TiO}_2$  nanostructures with immobilized BLV antigens before and after interaction with BLV antibodies, correspondently.

The changes of peak position after adsorption of Ab were calculated according following equation:

$$\Delta\lambda = \lambda_{\text{Ag-BSA}} - \lambda_{\text{Ab}}, \quad (22.2)$$

where  $\lambda_{\text{Ag-BSA}}$  and  $\lambda_{\text{Ab}}$  are PL peak positions of  $\text{TiO}_2$  nanostructures with immobilized BLV antigens before and after interaction with BLV antibodies, correspondently.

The results, obtained with the use of equations (22.1) and (22.2) are plotted in Fig. 22.4a. The analysis of the results showed that the changes of biosensor parameters had similar behavior. The obtained experimental curves increased at the range of Ab concentrations from 2–10 mg/ml. The further increase of Ab concentration led to the saturation of signal changes.

It is known that photoluminescence spectrum of  $\text{TiO}_2$  nanostructures can be split into two peaks, related to self-trapped excitons (STE) and oxygen vacancies  $V_{[\text{O}]}$ . We performed the fitting of the obtained PL spectra before and after interaction with biological molecules (the fitting is not shown here) and plotted the ratio of PL intensities related to STE and oxygen vacancies (Fig. 22.4b). It was found that the positions of both peaks have not been changed for all steps of the experiment (1–1.5 nm). At the same time the redistribution of the integrated intensity between peaks caused by STE at 461 nm and oxygen vacancies  $V_{[\text{O}]}$  at 502 nm was observed.

The formation of biosensitive layer and BSA molecules adsorption was accompanied by an intensity increase of the peak at 461 nm and a decrease of the peak at 502 nm, leading to a shift in the overall spectrum to shorter wavelengths. This indicates that the biomolecules adsorption by the surface of  $\text{TiO}_2$  reduces

the rate of radiative recombination caused by oxygen vacancies and increases the photoluminescence intensity caused by STE. After immune reaction between antigens and antibodies, the ratio of PL intensities related to STE and oxygen vacancies decreased with increasing concentration of antibodies.

The formation of antibody-antigen complex can lead to the changes in molecular structure of antigens previously immobilized on TiO<sub>2</sub> surface which is accompanied by link changes between antigens and TiO<sub>2</sub> surface.

Successful results were obtained for Salmonella antigens detection using similar procedure and biosensor, based on photoluminescence from ZnO nanorods [29–31]. The studied ZnO nanorods can be used as transducers in optical biosensors for Salmonella detection. The optimal response of the fabricated biosensor is observed at concentrations 10<sup>2</sup>–10<sup>6</sup> cells/ml.

## 22.4 Mechanism of Interaction Between TiO<sub>2</sub> Surface and Bio-molecules

Reaction of TiO<sub>2</sub> with proteins (Ab, Ag and BSA) is due to non-covalent binding. Van der Waals and hydrophobic bonds are suggested as the mechanism of interaction between TiO<sub>2</sub> surface and BLV molecules. An increase of PL emission after biomolecules immobilization is observed as a result of charge transfer from bio-molecules to TiO<sub>2</sub> surface. The proteins are bound to the surface by several functional groups affecting the surface band bending. The changes in UV PL emission after target molecules adsorption could occur due to the immune reaction between Ag-Ab couple. The decrease in intensity of the PL peak after target molecules adsorption is due to elimination and/or weakening of link between TiO<sub>2</sub> and bio-molecules, caused by structural modification of previously adsorbed Ab molecules as a result of antigen-antibody reaction. However, the mechanism of interaction cannot be determined from only measurements of PL. Additional methods, such as XPS confocal microscopy are needed to investigate it.

## 22.5 Conclusions

Specific selective interaction between immobilized antibodies and antigens couples ('key'-'lock' principle) can be monitored by PL of TiO<sub>2</sub> and ZnO. Photoluminescence from these nanomaterials is a suitable method for characterizing sample surface, surface defects and the changes of the surface properties as a result of biological impact. TiO<sub>2</sub> and ZnO nanostructures can be successfully used as a platform for the immobilization of biologically active substances on their surface which is confirmed by the changes in PL intensity and PL peak shift. Photoluminescence method of analyte detection is not the highest one comparably to SERS, SPR or

fluorescence however it is easier to be applied, does not need label system and can become the next generation of sensing devices. Obtained results provide a basis for the prospective application of metal oxide based nanostructures in immune biosensors for rapid diagnosis of such viruses as Bovine leucosis, Salmonella and other dangerous biological species.

## References

1. Ming-Chung Wu, Sápi A, Avila A, Szabó M, Hiltunen J, Huuhtanen M, Tóth G, Kukovecz Á, Kónya Z, Keiski R, Su W-F, Jantunen H, Kordás K (2011) Enhanced photocatalytic activity of TiO<sub>2</sub> nanofibers and their flexible composite films: decomposition of organic dyes and efficient H<sub>2</sub> generation from ethanol–water mixtures. *Nano Res* 4(4):360–369
2. Comini E, Baratto C, Faglia G, Ferroni M, Vomiero A, Sberveglieri G (2009) Quasi-one dimensional metal oxide semiconductors: preparation, characterization and application as chemical sensors. *Prog Mater Sci* 54:1–67
3. Alla T, Mikhael B, Roman V, Volodymyr K, Valentyn S, Nikolay S, Rositza Y (2016) Optical biosensors based on ZnO nanostructures: advantages and perspectives. A review. *Sensors and Actuators B* 229:664–677
4. Qiu J, Zhang Sh, Zhao H (2011) Recent applications of TiO<sub>2</sub> nanomaterials in chemical sensing in aqueous media. *Sens Actuators B* 160:875–890
5. Plugaru R, Cremades A, Piqueras J (2004) The effect of annealing in different atmospheres on the luminescence of polycrystalline TiO<sub>2</sub>. *J Phys Condens Matter* 16:S261–S268. PII: S0953-8984(04), 67008-1
6. Holford TRJ, Davis F, Higson SPJ (2012) Recent trends in antibody based sensors. *Biosens Bioelectron* 34:12–24
7. Viswanathan S, Rani Ch, Ja-an Annie Ho (2012) Electrochemical immunosensor for multiplexed detection of food-borne pathogens using nanocrystal bioconjugates and MWCNT screen-printed electrode. *Talanta* 94:315–319
8. Brandão D, Liébana S, Campoy S, Cortés P, Alegret S, Pividori MI (2013) Electrochemical magneto-immunosensing of Salmonella based on nano and micro-sized magnetic particles. *J Phys Conf Ser* 421:012020. doi:10.1088/1742-6596/421/1/012020
9. Stevanovic A, Büttner M, Zhang Zh, Yates JT Jr (2012) Photoluminescence of TiO<sub>2</sub>: effect of UV light and adsorbed molecules on surface band structure. *J Am Chem Soc* 134: 324–332
10. Drbohlavova J, Chomoucka J, Hrdy R, Prasek J, Janu L, Ryvolova M, Adam V, Kizek R, Halasova T, Hubalek J (2012) Effect of nucleic acid and albumin on luminescence properties of deposited TiO<sub>2</sub> quantum dots. *Int J Electrochem Sci* 7:1424–1432
11. Si P, Ding S, Yuan J, Lou XW, Kim DH (2011) Hierarchically structured one-dimensional TiO<sub>2</sub> for protein immobilization, direct electrochemistry, and mediator-free glucose sensing. *ACS Nano* 5(9):7617–7626
12. Setaro A, Lettieri S, Diamare D, Maddalena P, Malagù C, Carotta MC, Martinelli G (2008) Nanograined anatase titania-based optochemical gas detection. *New J Phys* 10:053030
13. Mun K-Sh, Alvarez SD, Choi W-Yo, Sailor MJ (2010) A stable, label-free optical interferometric biosensor based on TiO<sub>2</sub>. *ACS NANO* 4(4):2070–2076
14. Rodionov VE, Shnidko IN, Zolotovskiy A, Kruchinin SP (2013) Electroluminescence of Y<sub>2</sub>O<sub>3</sub>:Eu and Y<sub>2</sub>O<sub>3</sub>:Sm films. *Mater Sci* 31:232–239
15. Balida V, Ferrer JF (1977) Expression of the bovine leukemia virus and its internal antigen in blood lymphocytes. *Proc Soc Exp Biol Med* 156:388–391
16. Matthews R (1979) Classification and nomenclature of viruses. *Int Virol* 12:128–296

17. Wu M-Ch, Sápi A, Avila A, Szabó M, Hiltunen J, Huuhtanen M, Tóth G, Kukovecz Á, Kónya Z, Keiski R, Su W-F, Jantunen H, Kordás K (2011) Enhanced photocatalytic activity of TiO<sub>2</sub> nanofibers and their flexible composite films: decomposition of organic dyes and efficient H<sub>2</sub> generation from ethanol–water mixtures. *Nano Res* 4(4):360–369
18. Comini E, Baratto C, Faglia G, Ferroni M, Vomiero A, Sberveglieri G (2009) Quasi-one dimensional metal oxide semiconductors: preparation, characterization and application as chemical sensors. *Prog Mater Sci* 54:1–67
19. Chen X, Mao SS (2007) Titanium dioxide nanomaterials: synthesis, properties, modifications, and applications. *Chem Rev* 107(7):2891–2959
20. Qiu J, Zhang Sh, Zhao H (2011) Recent applications of TiO<sub>2</sub> nanomaterials in chemical sensing in aqueous media. *Sens Actuators B* 160:875–890
21. Plugaru R, Cremades A, Piqueras J (2004) The effect of annealing in different atmospheres on the luminescence of polycrystalline TiO<sub>2</sub>. *J Phys Condens Matter* 16:S261–S268. PII: S0953-8984(04) 67008-1
22. Preclíková J, Galá P, Trojánek F, Daniš S, Rezek B, Gregora I, Němcová Y, Malý P (2010) Nanocrystalline titanium dioxide films: influence of ambient conditions on surface- and volume-related photoluminescence. *J Appl Phys* 108:113502
23. Stevanovic A, Büttner M, Zhang Zh, Yates JT Jr (2012) Photoluminescence of TiO<sub>2</sub>: effect of UV light and adsorbed molecules on surface band structure. *J Am Chem Soc* 134:324–332
24. Drbohlavova J, Chomoucka J, Hrdy R, Prasek J, Janu L, Ryvolova M, Adam V, Kizek R, Halasova T, Hubalek J (2012) Effect of nucleic acid and albumin on luminescence properties of deposited TiO<sub>2</sub> quantum dots. *Int J Electrochem Sci* 7:1424–1432
25. Si P, Ding S, Yuan J, Lou XW, Kim DH (2011) Hierarchically structured one-dimensional TiO<sub>2</sub> for protein immobilization, direct electrochemistry, and mediator-free glucose sensing. *ACS Nano* 5(9):7617–7626
26. Mercado C, Seeley Z, Bandyopadhyay A, Bose S, McHale JL (2011) Photoluminescence of dense nanocrystalline titanium dioxide thin films: effect of doping and thickness and relation to gas sensing. *ACS Appl Mater Interfaces* 3:2281–2288
27. Nair PB, Justinivictor VB, Daniel GP, Joy K, Ramakrishnan V, Thomas PV (2011) Effect of RF power and sputtering pressure on the structural and optical properties of TiO<sub>2</sub> thin films prepared by RF magnetron sputtering. *Appl Surf Sci* 257:10869–10875
28. Li X, Gao C, Wang J, Lu B, Chen W, Song J, Zhang Sh, Zhang Zh, Pan X, Xie E (2012) TiO<sub>2</sub> films with rich bulk oxygen vacancies prepared by electrospinning for dye-sensitized solar cells. *J Power Sources* 214:244–250
29. Viter R et al (2012) Novel immune TiO<sub>2</sub> photoluminescence biosensors for leucosis detection. *Procedia Eng* 47:338–341
30. Smytyna V et al (2012) ZnO nanorods room temperature photoluminescence biosensors for salmonella detection. In: *Frontiers in optics, Rochester. Laser Science*, vol XXVIII, 14–18 Oct 2012
31. Viter R et al (2014) Application of room temperature photoluminescence from ZnO nanorods for salmonella detection. *IEEE Sens J* 14(6):2028–2034